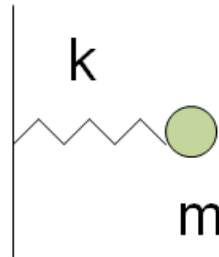
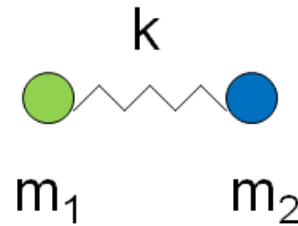


Classical harmonic oscillator

A Mass on a spring



B Diatomic

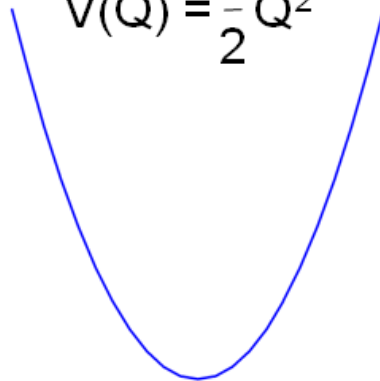


Reduced mass

$$\mu = \frac{m_1 m_2}{m_1 + m_2}$$

C Force constant k

$$V(Q) = \frac{k}{2} Q^2$$



D

$$\frac{\mu}{2} \left(\frac{\partial Q}{\partial t} \right)^2 + \frac{k}{2} Q^2 = 0$$

The classical harmonic oscillator obeys a Hooke's law equation:

$$F = -kQ$$

where k is a restoring force. A trial solution is:

$$Q(t) = Q_0 \cos(\omega t)$$

When substituted into the Hooke's law equation:

$$-\mu\omega^2 Q_0 \cos(\omega t) = -kQ_0 \cos(\omega t)$$

We can solve for the natural frequency of the spring
And we can also express that in cm^{-1} .

$$\omega = \sqrt{\frac{k}{\mu}} \quad \tilde{\nu}(\text{cm}^{-1}) = \frac{1}{2\pi c} \sqrt{\frac{k}{\mu}}$$

Harmonic approximation

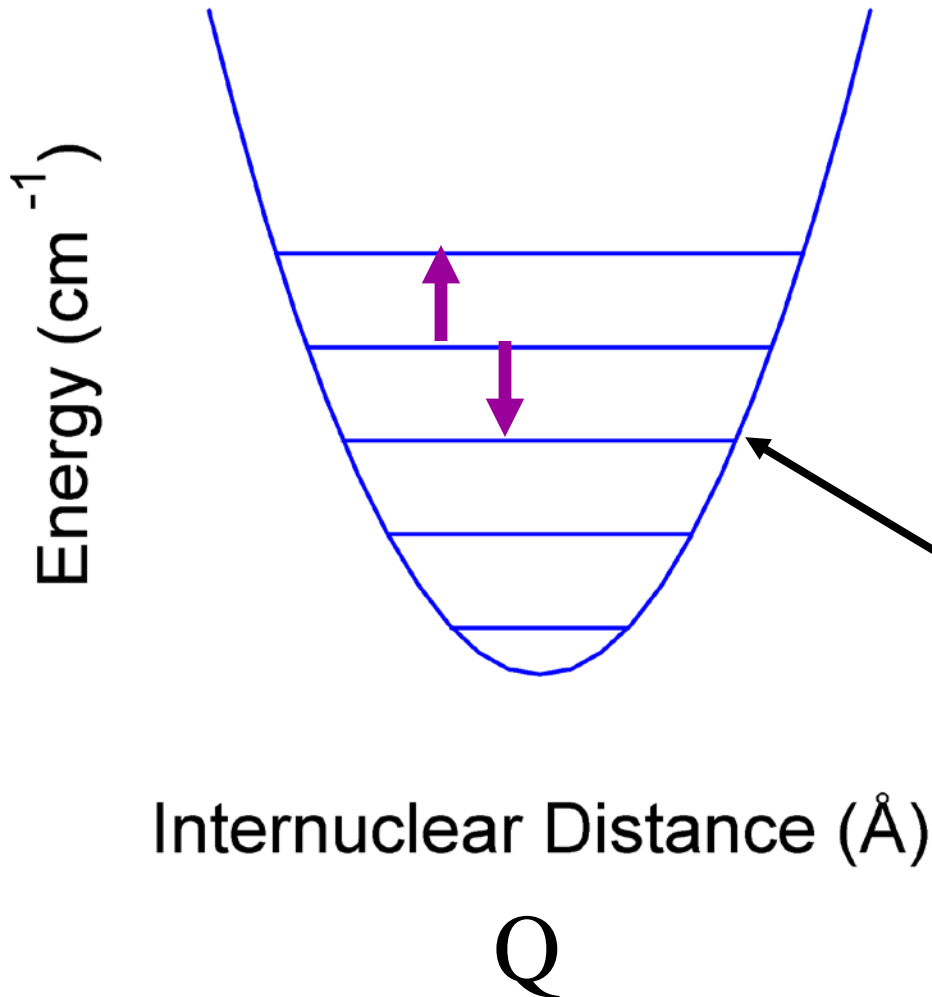
$$V(Q) = V(Q_0) + \left(\frac{\partial V}{\partial Q}\right) (Q - Q_0) + \frac{1}{2} \left(\frac{\partial^2 V}{\partial Q^2}\right) (Q - Q_0)^2 + \dots$$

At equilibrium $\left(\frac{\partial V}{\partial Q}\right) = 0$

Assume terms higher than quadratic are zero.
By definition

$$k = \left(\frac{\partial^2 V}{\partial Q^2}\right)$$

Quantum approach to the vibrational harmonic oscillator



Solution is Gaussian
Energy is quantized

$$-\frac{\hbar^2}{2\mu} \frac{\partial^2}{\partial Q^2} \chi + \frac{k}{2} Q^2 \chi = E \chi$$

$$E = \left(\nu + \frac{1}{2} \right) h\nu$$

ν is the quantum number

Allowed transitions

$$\nu' \rightarrow \nu + 1, \nu' \rightarrow \nu - 1$$

We can use a harmonic potential in the Schrödinger equation to calculate the wave functions and energies of the vibrations of molecules.

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$$y = \sqrt{\alpha} Q \quad \text{where } \alpha = \frac{\mu\omega}{\hbar} \quad \text{and } \epsilon = \frac{2E}{\hbar\omega}$$

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Noting that

$$\frac{\partial^2}{\partial Q^2} = \frac{\mu\omega}{\hbar} \frac{\partial^2}{\partial y^2}$$

we can write the equation as

$$-\frac{\partial^2}{\partial y^2} \chi + y^2 \chi = \epsilon \chi$$

One approach to solving such an equation is to find an asymptotic solution $g(y)$ assuming that $\varepsilon \sim 0$. Then, we can assume that the true solution is the product of $g(y)$ and a function $f(y)$. The asymptotic solution is:

$$\frac{\partial^2}{\partial y^2} \chi \approx y^2 \chi$$

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$$\frac{\partial^2}{\partial y^2} \chi \approx y^2 \chi$$

$f(y)$ can be a series expansion that will give different solutions for various values of ε . A Gaussian function is an appropriate trial solution for the this equation,

$$\chi_{trial} = e^{-y^2/2} \qquad \frac{\partial^2}{\partial y^2} \chi_{trial} = (y^2 - 1)e^{-y^2/2}$$

For large values of y we have

$$\frac{\partial^2}{\partial y^2} \chi_{trial} \approx y^2 e^{-y^2/2}$$

Thus, our trial solution for the general equation is

$$\chi_{trial} = f(y)e^{-y^2/2}$$

Substitution of the trial solution

In order to substitute this equation we need the derivatives.

We have

$$\frac{\partial \chi_{trial}}{\partial y} = \left(\frac{\partial f}{\partial y} - fy \right) e^{-y^2/2}$$

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Substituting this into the above equation gives us

$$\frac{\partial^2 f}{\partial y^2} - 2y \frac{\partial f}{\partial y} + (\epsilon - 1)f = 0$$

Frobenius series

If we assume that $f(y)$ has the form of a series

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$$\frac{\partial^2 f}{\partial y^2} = \sum_{n=0}^{\infty} n(n-1) a_n y^{n-2} = \sum_{n=0}^{\infty} (n+1)(n+2) a_{n+2} y^n$$

Series solution of the equation

$$\sum_{n=0}^{\infty} (n+1)(n+2)a_{n+2}y^n - 2y \sum_{n=0}^{\infty} na_ny^{n-1} + (\epsilon - 1) \sum_{n=0}^{\infty} a_ny^n = 0$$

$$\sum_{n=0}^{\infty} ((n+1)(n+2)a_{n+2} + (\epsilon - 1 - 2n)a_n)y^n = 0$$

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Once we choose a value for ϵ there is one and only one sequence of coefficients, a_n that defines the function $f(y)$.

Therefore, the sum can be zero for all values of y if and only if the coefficient of each power of

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$$(n+1)(n+2)a_{n+2} + (\epsilon - 1 - 2n)a_n = 0$$

And y vanishes separately. Thus,

$$a_{n+2} = \frac{1 + 2n - \epsilon}{(n+1)(n+2)} a_n$$

Energies of the quantum oscillator

Rather than finding an infinite series (which would actually be divergent in this case!) we will assume that the solution is a polynomial that terminates after a finite number of terms, n . The condition for the series to terminate is

$$a_{n+2} = 0$$

or

$$1 + 2n - \epsilon = 0$$

which implies

$$\epsilon = 2n + 1$$

Therefore, from the above we have

$$E = \frac{1}{2}(2n + 1)\hbar\omega = \left(n + \frac{1}{2}\right)\hbar\omega$$

Wave functions of the quantum harmonic oscillator

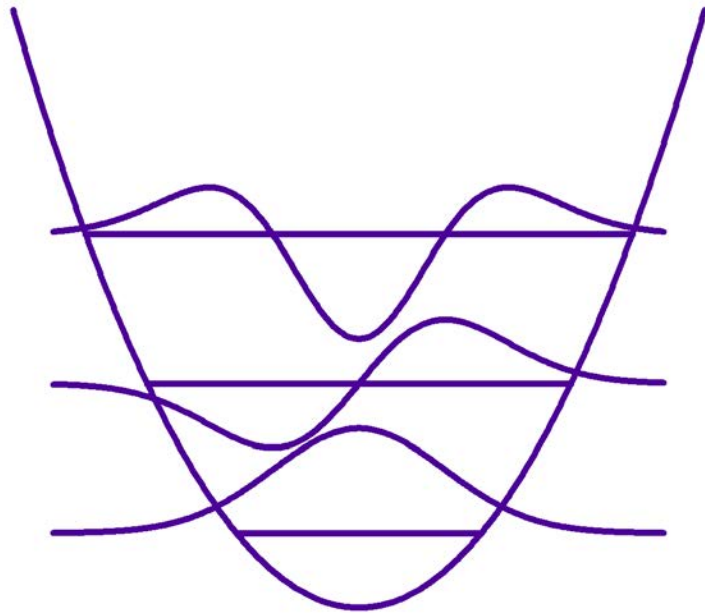
Using the definition of α , the solutions have the form:

$$\chi_0 = \left(\frac{\alpha}{\pi}\right)^{1/4} e^{-\alpha Q^2/2}$$

$$\chi_1 = \left(\frac{\alpha}{\pi}\right)^{1/4} \sqrt{2\alpha} Q e^{-\frac{\alpha Q^2}{2}}$$

$$\chi_2 = \left(\frac{\alpha}{\pi}\right)^{1/4} \frac{(4\alpha Q^2 - 2)}{2\sqrt{2}} e^{-\alpha Q^2/2}$$

Vibrational wavefunctions and energies



- Energy levels are given by
$$E_v = (v + 1/2)\hbar\omega$$
- Typical energies are of the order of 0 - 3200 cm^{-1}
- Wavefunctions are
$$\Psi_v = N_v H_v e^{-y^2/2}$$
where H_v is the Hermite polynomial

Solutions to harmonic oscillator

The Hermite polynomials are derivatives of a Gaussian

$$y = \sqrt{\alpha}Q \quad \text{where } \alpha = \frac{\mu\omega}{\hbar}$$

The Hermite generating function is

$$H_\nu(y) = (-1)^\nu e^{y^2/2} \frac{d^\nu}{dy^\nu} e^{-y^2/2}$$

The normalization constant is

$$N_\nu = \frac{1}{\sqrt{\alpha\pi^{1/2} 2^\nu \nu!}}$$

Hermite polynomials

$$\nu \quad H_\nu(y)$$

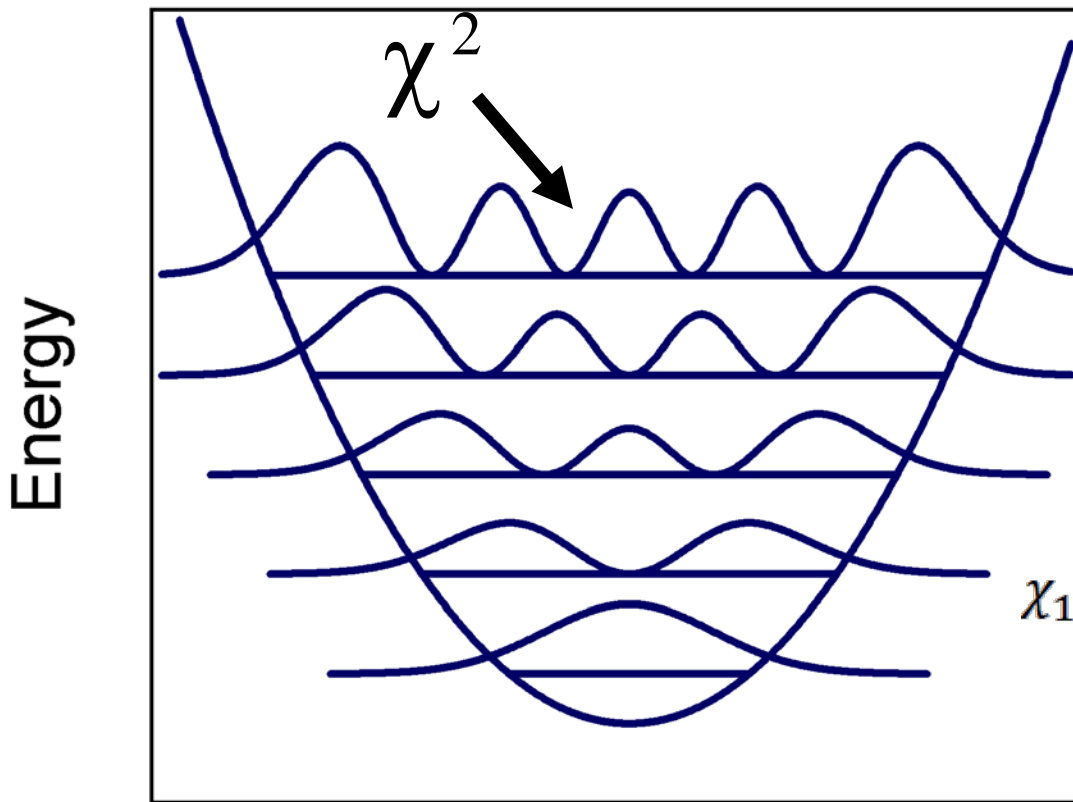
$$0 \quad 1$$

$$1 \quad 2y$$

$$2 \quad 4y^2 - 2$$

$$3 \quad 8y^3 - 12y$$

The square of the wave function gives rise to the probability distribution



The probability is shown in the figure.

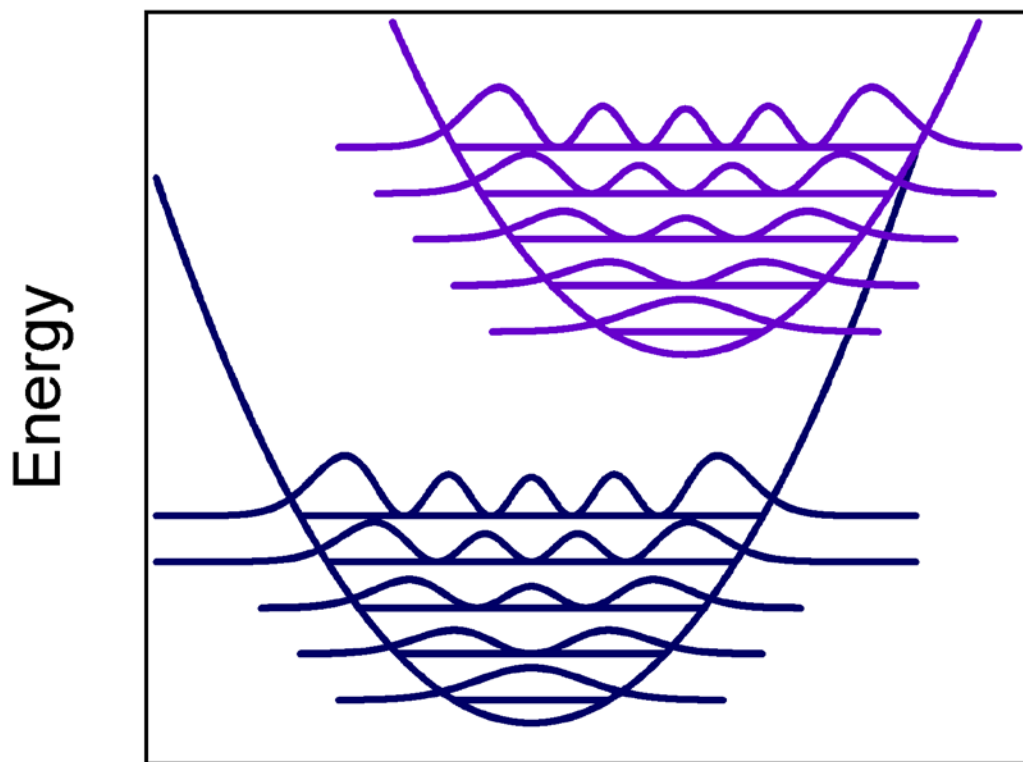
Solutions are Gaussians multiplied by polynomial functions.

$$\chi_1 = \left(\frac{\alpha}{\pi}\right)^{1/2} \sqrt{2\alpha} Q \exp\{-\alpha Q^2/2\}$$

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Nuclear Displacement

There is a potential energy surface that corresponds to each electronic state of the molecule



Nuclear Displacement

The shift in the nuclear displacement arises from the fact that the bond length increases in the σ^* state compared to the σ state. We will show that the overlap of the vibrational wave functions is key to understanding the shape of absorption bands.