A free electron has wave function  $\Psi(x, t) = \sin(kx - \omega t)$ . Determine the electron's de Broglie wavelength (in Å), momentum (in MKS), kinetic energy (in eV) and speed (in m/s) when k = 50 nm<sup>-1</sup> and the particle's mass is m = 9.1 x 10<sup>-31</sup> kg.

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The wavelength is given by:

$$k = \frac{2\pi}{\lambda}$$

Therefore:

$$\lambda = \frac{2\pi}{k} = \frac{2\pi}{50 \ x \ 10^9 \ m^{-1}} = 1.26 \ x \ 10^{-10} \ m = 1.26 \ \text{\AA}$$

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The momentum is given by:

$$p = \hbar k$$

So that

$$p = (1.054 x \, 10^{-34} \, Js)(50 \, x 10^9 \, m^{-1})$$

 $p = 5.26 x \, 10^{-24} \, kg \, m/s$ 

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The energy is given by:

$$E = \frac{\hbar^2 k^2}{2m}$$

So that

$$E = \frac{\hbar^2 k^2}{2m} = \frac{(1.054 \ x \ 10^{-34} \ Js)^2 (50 \ x \ 10^9 \ m^{-1})^2}{2(9.1 \ x \ 10^{-31} \ kg)}$$

 $E = 1.53 x \, 10^{-17} J$  which is

$$\frac{1.53 \ x \ 10^{-17} \ J}{1.602 \ x \ 10^{-19} \ C} = 95.2 \ eV$$

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The velocity is given by:

$$v = \frac{p}{m} = \frac{\hbar k}{m}$$

So that

$$v = \frac{5.26 \, x \, 10^{-24} \, kg \, m/s}{9.1 \, x 10^{-31} \, kg}$$

 $v = 5.78 x 10^6 m/s$