## Number of moles of gas

How many moles are in 2000 L of the atmosphere at 373 K?

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## Solution:

$$
\begin{gathered}
n=\frac{P V}{R T}=\frac{(1 \mathrm{~atm})(2000 \mathrm{~L})}{(0.08206 \mathrm{Latm} / \mathrm{molK})(373 \mathrm{~K})} \\
n=65.3 \mathrm{moles}
\end{gathered}
$$

