

## **Cold Pack Temperature**

A cold pack consists of NaNO<sub>3</sub>. When the pack breaks open the salt is released into water and causes cooling because of the endothermic reaction. Given the enthalpy and entropy data below, what is

Given the enthalpy and entropy data below, what is the minimum temperature at which NaNO<sub>3</sub> will dissolve in water?

$$\Delta_{solv}H^o(NaNO_3) = +42.5 \ kJ/mol$$

$$\Delta_{solv}S^{o}(NaNO_3) = +165 J/molK$$