



# Heating of a fuel in an engine

Consider the fact that after combustion the octane fuel has been converted into  $\text{CO}_2$  and  $\text{H}_2\text{O}$  in the vapor phase. Assuming an average value of  $c_p = 33 \text{ J/mol-K}$  for vapor produced by combustion, what is the final temperature if 12 microliters of octane are combusted? ( $\rho_{\text{oct}} = 0.7 \text{ gm/cm}^3$ )