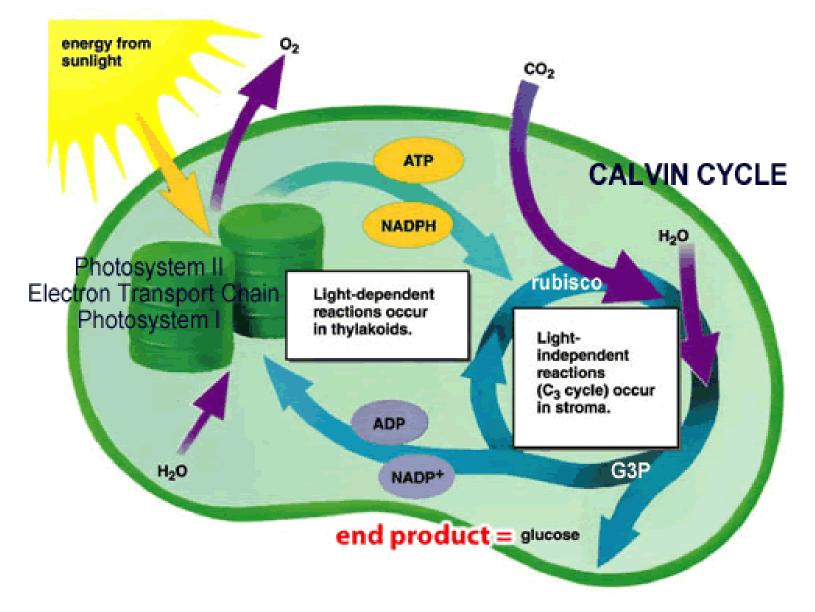
Theoretical maximum glucose production



Each year plants fix 1.4 billion tons of CO_2 . $CO_2 + H_2O = C_6H_{12}O_6 + O_2$ Given that the concentration of CO_2 is 380 ppm and the pressure of CO_2 is 3.8 x 10⁻⁴ atm at 298 K. Given that the mass of the atmosphere is 5 x 10¹⁸ kg, calculate the mass of carbon that would be fixed as glucose if all of the CO_2 were taken up by plants.