

Reaction Rates

The reaction rate for the following was 0.03 M/s



What is the rate of change for each component?

Rates for individual reactants

The reaction rate for the following was 0.03 M/s



What is the rate of change for each component?

$$d[\text{N}_2\text{O}_5]/dt = -0.03 \text{ M/s}$$

$$d[\text{NO}_2]/dt = 0.06 \text{ M/s}$$

$$d[\text{O}_2]/dt = 0.015 \text{ M/s}$$