## Balancing Chemical Equations

## Balance the chemical reaction:

$$
\ldots \mathrm{PCl}_{5}+\ldots \mathrm{H}_{2} \mathrm{O} \rightarrow \ldots \mathrm{H}_{3} \mathrm{PO}_{4}+\ldots \ldots \mathrm{HCl}
$$

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Step 1. Write out coefficients

$$
\mathrm{a} \mathrm{PCl}_{5}+\mathrm{bH}_{2} \mathrm{O} \rightarrow \mathrm{xH}_{3} \mathrm{PO}_{4}+\mathrm{y} \mathrm{HCl}
$$

Step 2. Construct the atom equations:
P: $a=x$
$\mathrm{Cl}: 5 \mathrm{a}=\mathrm{y}$
$H: 2 b=3 x+y$
$0: b=4 x$

## Balancing Chemical Equations

Step 3. Make an initial guess and solve for coefficients:
P: $a=x$
Cl: $5 \mathrm{a}=\mathrm{y} \operatorname{Try} \mathrm{a}=1$, then $\mathrm{y}=5$
$H: 2 b=3 x+y$
$0: b=4 x$

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Step 3. Make an initial guess and solve for coefficients:
$\mathrm{P}: \mathrm{a}=\mathrm{x}$ and $\mathrm{x}=1$
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Cl: $5 \mathrm{a}=\mathrm{y} \operatorname{Try} \mathrm{a}=1$, then $\mathrm{y}=5$
$\mathrm{H}: 2 \mathrm{~b}=3 \mathrm{x}+\mathrm{y}$ Use this equation to check consistency!
$0: b=4 x$ and $b=4$

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Step 3. Make an initial guess and solve for coefficients:
$\mathrm{P}: \mathrm{a}=\mathrm{x}$ and $\mathrm{x}=1$
Cl: $5 \mathrm{a}=\mathrm{y} \operatorname{Try} \mathrm{a}=1$, then $\mathrm{y}=5$
$\mathrm{H}: 2 \mathrm{~b}=3 \mathrm{x}+\mathrm{y}$ Use this equation to check consistency!
$0: b=4 x$ and $b=4$

$$
2(4)=3(1)+5 \text { ? Yes! }
$$

Write the balanced reaction:

$$
\mathrm{PCl}_{5}+4 \mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{H}_{3} \mathrm{PO}_{4}+5 \mathrm{HCl}
$$

