Balance the chemical reaction:

$$___PCl_5 + ___H_2O \rightarrow ___H_3PO_4 + ___HCl$$

Balance the chemical reaction:

$$___$$
 PCl₅ + $___$ H₂O \rightarrow $___$ H₃PO₄ + $___$ HCl

Step 1. Write out coefficients

a
$$PCl_5 + b H_2O \rightarrow x H_3PO_4 + y HCl$$

Step 2. Construct the atom equations:

P: a = x

Cl: 5a = y

H: 2b = 3x + y

O: b = 4x

Step 3. Make an initial guess and solve for coefficients:

$$P: a = x$$

Cl:
$$5a = y \text{ Try } a = 1$$
, then $y = 5$

H:
$$2b = 3x + y$$

O:
$$b = 4x$$

Step 3. Make an initial guess and solve for coefficients:

P:
$$a = x \text{ and } x = 1$$

Cl:
$$5a = y \text{ Try } a = 1$$
, then $y = 5$

H:
$$2b = 3x + y$$

O:
$$b = 4x$$

Step 3. Make an initial guess and solve for coefficients:

P: a = x and x = 1

Cl: 5a = y Try a = 1, then y = 5

H: 2b = 3x + y Use this equation to check consistency!

O: b = 4x and b = 4

Step 3. Make an initial guess and solve for coefficients:

P:
$$a = x \text{ and } x = 1$$

Cl:
$$5a = y \text{ Try } a = 1$$
, then $y = 5$

H: 2b = 3x + y Use this equation to check consistency!

O:
$$b = 4x \text{ and } b = 4$$

$$2(4) = 3(1) + 5$$
? Yes!

Write the balanced reaction:

$$PCl_5 + 4 H_2O \rightarrow H_3PO_4 + 5 HCl$$