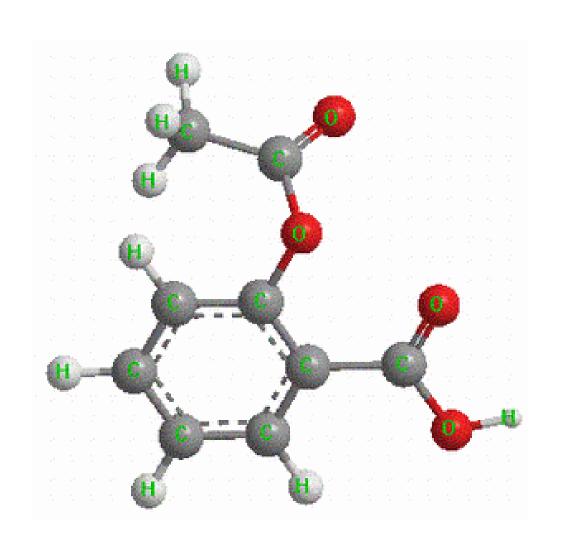
## Molarity of aspirin



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We can consider some rudimentary pharmacology using our knowledge of molar mass of aspirin. Aspirin has a chemical formula  $C_9O_4H_8$ . Assuming that a grown person takes 1 gram of aspirin and that all of the aspirin is taken up into the blood stream (total volume is 5 liters), then what is the concentration of aspirin in the blood (in units of molarity)?